

1. Hydrogen (H)

- **Atomic Number:** 1 $\rightarrow$ 00000001
- **Configuration:** s-block (Group 1) $\rightarrow$ 1000
- **Valence Electrons:** 1 $\rightarrow$ 001
- **Electronegativity:** $2.20 < 2.5 \rightarrow 0$
- **Period:** 1 $\rightarrow$ 001
- **Group:** 1 $\rightarrow$ 00001
- **Vector:** 00000001 1000 001 0 00100001

2. Helium (He)

- **Atomic Number:** 2 $\rightarrow$ 00000010
- **Configuration:** s-block (Group 18) $\rightarrow$ 1000
- **Valence Electrons:** Noble gas, full shell $\rightarrow$ 000
- **Electronegativity:** Undefined (noble gas), set to 0 $\rightarrow$ 0
- **Period:** 1 $\rightarrow$ 001
- **Group:** 18 $\rightarrow$ 10010
- **Vector:** 00000010 1000 000 0 00110010

3. Lithium (Li)

- **Atomic Number:** 3 $\rightarrow$ 00000011
- **Configuration:** s-block (Group 1) $\rightarrow$ 1000
- **Valence Electrons:** 1 $\rightarrow$ 001
- **Electronegativity:** $0.98 < 2.5 \rightarrow 0$
- **Period:** 2 $\rightarrow$ 010
- **Group:** 1 $\rightarrow$ 00001
- **Vector:** 00000011 1000 001 0 01000001

4. Beryllium (Be)

- **Atomic Number:** 4 $\rightarrow$ 00000100
- **Configuration:** s-block (Group 2) $\rightarrow$ 1000
- **Valence Electrons:** 2 $\rightarrow$ 010

- **Electronegativity:** $1.57 < 2.5 \rightarrow 0$
- **Period:** 2 $\rightarrow 010$
- **Group:** 2 $\rightarrow 00010$
- **Vector:** 00000100 1000 010 0 01000010

5. Boron (B)

- **Atomic Number:** 5 $\rightarrow 00000101$
- **Configuration:** p-block (Group 13) $\rightarrow 1100$
- **Valence Electrons:** $13 - 10 = 3 \rightarrow 011$
- **Electronegativity:** $2.04 < 2.5 \rightarrow 0$
- **Period:** 2 $\rightarrow 010$
- **Group:** 13 $\rightarrow 01101$
- **Vector:** 00000101 1100 011 0 01001101

6. Carbon (C)

- **Atomic Number:** 6 $\rightarrow 00000110$
- **Configuration:** p-block (Group 14) $\rightarrow 1100$
- **Valence Electrons:** $14 - 10 = 4 \rightarrow 100$
- **Electronegativity:** $2.55 \geq 2.5 \rightarrow 1$
- **Period:** 2 $\rightarrow 010$
- **Group:** 14 $\rightarrow 01110$
- **Vector:** 00000110 1100 100 1 01001110

7. Nitrogen (N)

- **Atomic Number:** 7 $\rightarrow 00000111$
- **Configuration:** p-block (Group 15) $\rightarrow 1100$
- **Valence Electrons:** $15 - 10 = 5 \rightarrow 101$
- **Electronegativity:** $3.04 \geq 2.5 \rightarrow 1$
- **Period:** 2 $\rightarrow 010$
- **Group:** 15 $\rightarrow 01111$
- **Vector:** 00000111 1100 101 1 01001111

8. Oxygen (O)

- **Atomic Number:** 8 → 00001000
- **Configuration:** p-block (Group 16) → 1100
- **Valence Electrons:** $16 - 10 = 6$ → 110
- **Electronegativity:** $3.44 \geq 2.5$ → 1
- **Period:** 2 → 010
- **Group:** 16 → 10000
- **Vector:** 00001000 1100 110 1 01010000

9. Fluorine (F)

- **Atomic Number:** 9 → 00001001
- **Configuration:** p-block (Group 17) → 1100
- **Valence Electrons:** $17 - 10 = 7$ → 111
- **Electronegativity:** $3.98 \geq 2.5$ → 1
- **Period:** 2 → 010
- **Group:** 17 → 10001
- **Vector:** 00001001 1100 111 1 01010001

10. Neon (Ne)

- **Atomic Number:** 10 → 00001010
- **Configuration:** p-block (Group 18) → 1100
- **Valence Electrons:** Noble gas, full shell → 000
- **Electronegativity:** Undefined, set to 0 → 0
- **Period:** 2 → 010
- **Group:** 18 → 10010
- **Vector:** 00001010 1100 000 0 01010010

11. Sodium (Na)

- **Atomic Number:** 11 → 00001011
- **Configuration:** s-block (Group 1) → 1000

- **Valence Electrons:** 1 $\rightarrow$ 001
- **Electronegativity:** 0.93 < 2.5 $\rightarrow$ 0
- **Period:** 3 $\rightarrow$ 011
- **Group:** 1 $\rightarrow$ 00001
- **Vector:** 00001011 1000 001 0 011 00001

12. Magnesium (Mg)

- **Atomic Number:** 12 $\rightarrow$ 00001100
- **Configuration:** s-block (Group 2) $\rightarrow$ 1000
- **Valence Electrons:** 2 $\rightarrow$ 010
- **Electronegativity:** 1.31 < 2.5 $\rightarrow$ 0
- **Period:** 3 $\rightarrow$ 011
- **Group:** 2 $\rightarrow$ 00010
- **Vector:** 00001100 1000 010 0 011 00010

13. Aluminum (Al)

- **Atomic Number:** 13 $\rightarrow$ 00001101
- **Configuration:** p-block (Group 13) $\rightarrow$ 1100
- **Valence Electrons:** 3 $\rightarrow$ 011
- **Electronegativity:** 1.61 < 2.5 $\rightarrow$ 0
- **Period:** 3 $\rightarrow$ 011
- **Group:** 13 $\rightarrow$ 01101
- **Vector:** 00001101 1100 011 0 011 01101

14. Silicon (Si)

- **Atomic Number:** 14 $\rightarrow$ 00001110
- **Configuration:** p-block (Group 14) $\rightarrow$ 1100
- **Valence Electrons:** 4 $\rightarrow$ 100
- **Electronegativity:** 1.9 < 2.5 $\rightarrow$ 0
- **Period:** 3 $\rightarrow$ 011
- **Group:** 14 $\rightarrow$ 01110
- **Vector:** 00001110 1100 100 0 011 01110

15. Phosphorus (P)

- **Atomic Number:** 15 $\rightarrow$ 00001111
- **Configuration:** p-block (Group 15) $\rightarrow$ 1100
- **Valence Electrons:** 5 $\rightarrow$ 101
- **Electronegativity:** $2.19 < 2.5 \rightarrow 0$
- **Period:** 3 $\rightarrow$ 011
- **Group:** 15 $\rightarrow$ 01111
- **Vector:** 00001111 1100 101 0 011 01111

16. Sulfur (S)

- **Atomic Number:** 16 $\rightarrow$ 00010000
- **Configuration:** p-block (Group 16) $\rightarrow$ 1100
- **Valence Electrons:** 6 $\rightarrow$ 110
- **Electronegativity:** $2.58 \geq 2.5 \rightarrow 1$
- **Period:** 3 $\rightarrow$ 011
- **Group:** 16 $\rightarrow$ 10000
- **Vector:** 00010000 1100 110 1 011 10000

17. Chlorine (Cl)

- **Atomic Number:** 17 $\rightarrow$ 00010001
- **Configuration:** p-block (Group 17) $\rightarrow$ 1100
- **Valence Electrons:** 7 $\rightarrow$ 111
- **Electronegativity:** $3.16 \geq 2.5 \rightarrow 1$
- **Period:** 3 $\rightarrow$ 011
- **Group:** 17 $\rightarrow$ 10001
- **Vector:** 00010001 1100 111 1 011 10001

18. Argon (Ar)

- **Atomic Number:** 18 $\rightarrow$ 00010010
- **Configuration:** p-block (Group 18) $\rightarrow$ 1100

- **Valence Electrons:** Noble gas, full shell → 000
- **Electronegativity:** Undefined, set to 0 → 0
- **Period:** 3 → 011
- **Group:** 18 → 10010
- **Vector:** 00010010 1100 000 0 011 10010

19. Potassium (K)

- **Atomic Number:** 19 → 00010011
- **Configuration:** s-block (Group 1) → 1000
- **Valence Electrons:** 1 → 001
- **Electronegativity:** $0.82 < 2.5$ → 0
- **Period:** 4 → 100
- **Group:** 1 → 00001
- **Vector:** 00010011 1000 001 0 100 00001

20. Calcium (Ca)

- **Atomic Number:** 20 → 00010100
- **Configuration:** s-block (Group 2) → 1000
- **Valence Electrons:** 2 → 010
- **Electronegativity:** $1.0 < 2.5$ → 0
- **Period:** 4 → 100
- **Group:** 2 → 00010
- **Vector:** 00010100 1000 010 0 100 00010

21. Scandium (Sc)

- **Atomic Number:** 21 → 00010101
- **Configuration:** d-block (Group 3) → 1010
- **Valence Electrons:** 2 ($4s^2$) → 010
- **Electronegativity:** $1.36 < 2.5$ → 0
- **Period:** 4 → 100
- **Group:** 3 → 00011
- **Vector:** 00010101 1010 010 0 10000011

22. Titanium (Ti)

- **Atomic Number:** 22 → 00010110
- **Configuration:** d-block (Group 4) → 1010
- **Valence Electrons:** 2 ($4s^2$) → 010
- **Electronegativity:** $1.54 < 2.5$ → 0
- **Period:** 4 → 100
- **Group:** 4 → 00100
- **Vector:** 00010110 1010 010 0 100 00100

23. Vanadium (V)

- **Atomic Number:** 23 → 00010111
- **Configuration:** d-block (Group 5) → 1010
- **Valence Electrons:** 2 ($4s^2$) → 010
- **Electronegativity:** $1.63 < 2.5$ → 0
- **Period:** 4 → 100
- **Group:** 5 → 00101
- **Vector:** 00010111 1010 010 0 100 00101

24. Chromium (Cr)

- **Atomic Number:** 24 → 00011000
- **Configuration:** d-block (Group 6) → 1010
- **Valence Electrons:** 1 ($4s^1$ due to exception) → 001
- **Electronegativity:** $1.66 < 2.5$ → 0
- **Period:** 4 → 100
- **Group:** 6 → 00110
- **Vector:** 00011000 1010 001 0 100 00110

25. Manganese (Mn)

- **Atomic Number:** 25 → 00011001
- **Configuration:** d-block (Group 7) → 1010

- **Valence Electrons:** 2 ($4s^2$) $\rightarrow$ 010
- **Electronegativity:** $1.55 < 2.5 \rightarrow 0$
- **Period:** 4 $\rightarrow$ 100
- **Group:** 7 $\rightarrow$ 00111
- **Vector:** 00011001 1010 010 0 100 00111

26. Iron (Fe)

- **Atomic Number:** 26 $\rightarrow$ 00011010
- **Configuration:** d-block (Group 8) $\rightarrow$ 1010
- **Valence Electrons:** 2 ($4s^2$) $\rightarrow$ 010
- **Electronegativity:** $1.83 < 2.5 \rightarrow 0$
- **Period:** 4 $\rightarrow$ 100
- **Group:** 8 $\rightarrow$ 01000
- **Vector:** 00011010 1010 010 0 10001000

27. Cobalt (Co)

- **Atomic Number:** 27 $\rightarrow$ 00011011
- **Configuration:** d-block (Group 9) $\rightarrow$ 1010
- **Valence Electrons:** 2 ($4s^2$) $\rightarrow$ 010
- **Electronegativity:** $1.88 < 2.5 \rightarrow 0$
- **Period:** 4 $\rightarrow$ 100
- **Group:** 9 $\rightarrow$ 01001
- **Vector:** 00011011 1010 010 0 10001001

28. Nickel (Ni)

- **Atomic Number:** 28 $\rightarrow$ 00011100
- **Configuration:** d-block (Group 10) $\rightarrow$ 1010
- **Valence Electrons:** 2 ($4s^2$) $\rightarrow$ 010
- **Electronegativity:** $1.91 < 2.5 \rightarrow 0$
- **Period:** 4 $\rightarrow$ 100
- **Group:** 10 $\rightarrow$ 01010
- **Vector:** 00011100 1010 010 0 10001010

29. Copper (Cu)

- **Atomic Number:** 29 → 00011101
- **Configuration:** d-block (Group 11) → 1010
- **Valence Electrons:** 1 ($4s^1$ due to exception) → 001
- **Electronegativity:** $1.90 < 2.5$ → 0
- **Period:** 4 → 100
- **Group:** 11 → 01011
- **Vector:** 00011101 1010 001 0 10001011

30. Zinc (Zn)

- **Atomic Number:** 30 → 00011110
- **Configuration:** d-block (Group 12) → 1010
- **Valence Electrons:** 2 ($4s^2$) → 010
- **Electronegativity:** $1.65 < 2.5$ → 0
- **Period:** 4 → 100
- **Group:** 12 → 01100
- **Vector:** 00011110 1010 010 0 10001100

31. Gallium (Ga)

- **Atomic Number:** 31 → 00011111
- **Configuration:** p-block (Group 13) → 1100
- **Valence Electrons:** 3 → 011
- **Electronegativity:** $1.81 < 2.5$ → 0
- **Period:** 4 → 100
- **Group:** 13 → 01101
- **Vector:** 00011111 1100 011 0 10001101

32. Germanium (Ge)

- **Atomic Number:** 32 → 00100000
- **Configuration:** p-block (Group 14) → 1100

- **Valence Electrons:** 4 $\rightarrow$ 100
- **Electronegativity:** 2.01 < 2.5 $\rightarrow$ 0
- **Period:** 4 $\rightarrow$ 100
- **Group:** 14 $\rightarrow$ 01110
- **Vector:** 00100000 1100 100 0 10001110

33. Arsenic (As)

- **Atomic Number:** 33 $\rightarrow$ 00100001
- **Configuration:** p-block (Group 15) $\rightarrow$ 1100
- **Valence Electrons:** 5 $\rightarrow$ 101
- **Electronegativity:** 2.18 < 2.5 $\rightarrow$ 0
- **Period:** 4 $\rightarrow$ 100
- **Group:** 15 $\rightarrow$ 01111
- **Vector:** 00100001 1100 101 0 10001111

34. Selenium (Se)

- **Atomic Number:** 34 $\rightarrow$ 00100010
- **Configuration:** p-block (Group 16) $\rightarrow$ 1100
- **Valence Electrons:** 6 $\rightarrow$ 110
- **Electronegativity:** 2.55 $\geq$ 2.5 $\rightarrow$ 1
- **Period:** 4 $\rightarrow$ 100
- **Group:** 16 $\rightarrow$ 10000
- **Vector:** 00100010 1100 110 1 10010000

35. Bromine (Br)

- **Atomic Number:** 35 $\rightarrow$ 00100011
- **Configuration:** p-block (Group 17) $\rightarrow$ 1100
- **Valence Electrons:** 7 $\rightarrow$ 111
- **Electronegativity:** 2.96 $\geq$ 2.5 $\rightarrow$ 1
- **Period:** 4 $\rightarrow$ 100
- **Group:** 17 $\rightarrow$ 10001
- **Vector:** 00100011 1100 111 1 10010001

36. Krypton (Kr)

- **Atomic Number:** 36 → 00100100
- **Configuration:** p-block (Group 18) → 1100
- **Valence Electrons:** Noble gas, full shell → 000
- **Electronegativity:** Undefined, set to 0 → 0
- **Period:** 4 → 100
- **Group:** 18 → 10010
- **Vector:** 00100100 1100 000 0 10010010

37. Rubidium (Rb)

- **Atomic Number:** 37 → 00100101
- **Configuration:** s-block (Group 1) → 1000
- **Valence Electrons:** 1 → 001
- **Electronegativity:** $0.82 < 2.5$ → 0
- **Period:** 5 → 101
- **Group:** 1 → 00001
- **Vector:** 00100101 1000 001 0 10100001

38. Strontium (Sr)

- **Atomic Number:** 38 → 00100110
- **Configuration:** s-block (Group 2) → 1000
- **Valence Electrons:** 2 → 010
- **Electronegativity:** $0.95 < 2.5$ → 0
- **Period:** 5 → 101
- **Group:** 2 → 00010
- **Vector:** 00100110 1000 010 0 10100010

39. Yttrium (Y)

- **Atomic Number:** 39 → 00100111
- **Configuration:** d-block (Group 3) → 1010

- **Valence Electrons:** 2 $\rightarrow$ 010
- **Electronegativity:** 1.22 < 2.5 $\rightarrow$ 0
- **Period:** 5 $\rightarrow$ 101
- **Group:** 3 $\rightarrow$ 00011
- **Vector:** 00100111 1010 010 0 10100011

40. Zirconium (Zr)

- **Atomic Number:** 40 $\rightarrow$ 00101000
- **Configuration:** d-block (Group 4) $\rightarrow$ 1010
- **Valence Electrons:** 2 $\rightarrow$ 010
- **Electronegativity:** 1.33 < 2.5 $\rightarrow$ 0
- **Period:** 5 $\rightarrow$ 101
- **Group:** 4 $\rightarrow$ 00100
- **Vector:** 00101000 1010 010 0 101 00100

41. Niobium (Nb)

- **Atomic Number:** 41 $\rightarrow$ 00101001
- **Configuration:** d-block (Group 5) $\rightarrow$ 1010
- **Valence Electrons:** 2 ($5s^2$) $\rightarrow$ 010
- **Electronegativity:** 1.6 < 2.5 $\rightarrow$ 0
- **Period:** 5 $\rightarrow$ 101
- **Group:** 5 $\rightarrow$ 00101
- **Vector:** 00101001 1010 010 0 101 00101

42. Molybdenum (Mo)

- **Atomic Number:** 42 $\rightarrow$ 00101010
- **Configuration:** d-block (Group 6) $\rightarrow$ 1010
- **Valence Electrons:** 1 ($5s^1$ due to exception) $\rightarrow$ 001
- **Electronegativity:** 2.16 < 2.5 $\rightarrow$ 0
- **Period:** 5 $\rightarrow$ 101
- **Group:** 6 $\rightarrow$ 00110
- **Vector:** 00101010 1010 001 0 101 00110

43. Technetium (Tc)

- **Atomic Number:** 43 → 00101011
- **Configuration:** d-block (Group 7) → 1010
- **Valence Electrons:** 2 ($5s^2$) → 010
- **Electronegativity:** $1.9 < 2.5$ → 0
- **Period:** 5 → 101
- **Group:** 7 → 00111
- **Vector:** 00101011 1010 010 0 101 00111

44. Ruthenium (Ru)

- **Atomic Number:** 44 → 00101100
- **Configuration:** d-block (Group 8) → 1010
- **Valence Electrons:** 1 ($5s^1$ due to exception) → 001
- **Electronegativity:** $2.2 < 2.5$ → 0
- **Period:** 5 → 101
- **Group:** 8 → 01000
- **Vector:** 00101100 1010 001 0 101 01000

45. Rhodium (Rh)

- **Atomic Number:** 45 → 00101101
- **Configuration:** d-block (Group 9) → 1010
- **Valence Electrons:** 1 ($5s^1$ due to exception) → 001
- **Electronegativity:** $2.28 < 2.5$ → 0
- **Period:** 5 → 101
- **Group:** 9 → 01001
- **Vector:** 00101101 1010 001 0 101 01001

46. Palladium (Pd)

- **Atomic Number:** 46 → 00101110
- **Configuration:** d-block (Group 10) → 1010

- **Valence Electrons:** 0 (5s⁰, full 4d shell) → 000
- **Electronegativity:** 2.20 < 2.5 → 0
- **Period:** 5 → 101
- **Group:** 10 → 01010
- **Vector:** 00101110 1010 000 0 101 01010

47. Silver (Ag)

- **Atomic Number:** 47 → 00101111
- **Configuration:** d-block (Group 11) → 1010
- **Valence Electrons:** 1 (5s¹) → 001
- **Electronegativity:** 1.93 < 2.5 → 0
- **Period:** 5 → 101
- **Group:** 11 → 01011
- **Vector:** 00101111 1010 001 0 101 01011

48. Cadmium (Cd)

- **Atomic Number:** 48 → 00110000
- **Configuration:** d-block (Group 12) → 1010
- **Valence Electrons:** 2 (5s²) → 010
- **Electronegativity:** 1.69 < 2.5 → 0
- **Period:** 5 → 101
- **Group:** 12 → 01100
- **Vector:** 00110000 1010 010 0 101 01100

49. Indium (In)

- **Atomic Number:** 49 → 00110001
- **Configuration:** p-block (Group 13) → 1100
- **Valence Electrons:** 3 → 011
- **Electronegativity:** 1.78 < 2.5 → 0
- **Period:** 5 → 101
- **Group:** 13 → 01101
- **Vector:** 00110001 1100 011 0 101 01101

50. Tin (Sn)

- **Atomic Number:** 50 → 00110010
- **Configuration:** p-block (Group 14) → 1100
- **Valence Electrons:** 4 → 100
- **Electronegativity:** 1.96 < 2.5 → 0
- **Period:** 5 → 101
- **Group:** 14 → 01110
- **Vector:** 00110010 1100 100 0 101 01110

51. Antimony (Sb)

- **Atomic Number:** 51 → 00110011
- **Configuration:** p-block (Group 15) → 1100
- **Valence Electrons:** 5 → 101
- **Electronegativity:** 2.05 < 2.5 → 0
- **Period:** 5 → 101
- **Group:** 15 → 01111
- **Vector:** 00110011 1100 101 0 101 01111

52. Tellurium (Te)

- **Atomic Number:** 52 → 00110100
- **Configuration:** p-block (Group 16) → 1100
- **Valence Electrons:** 6 → 110
- **Electronegativity:** 2.1 < 2.5 → 0
- **Period:** 5 → 101
- **Group:** 16 → 10000
- **Vector:** 00110100 1100 110 0 101 10000

53. Iodine (I)

- **Atomic Number:** 53 → 00110101
- **Configuration:** p-block (Group 17) → 1100

- **Valence Electrons:** $7 \rightarrow 111$
- **Electronegativity:** $2.66 \geq 2.5 \rightarrow 1$
- **Period:** $5 \rightarrow 101$
- **Group:** $17 \rightarrow 10001$
- **Vector:** 00110101 1100 111 1 101 10001

54. Xenon (Xe)

- **Atomic Number:** $54 \rightarrow 00110110$
- **Configuration:** p-block (Group 18) $\rightarrow 1100$
- **Valence Electrons:** Noble gas, full shell $\rightarrow 000$
- **Electronegativity:** Undefined, set to 0 $\rightarrow 0$
- **Period:** $5 \rightarrow 101$
- **Group:** $18 \rightarrow 10010$
- **Vector:** 00110110 1100 000 0 101 10010

55. Cesium (Cs)

- **Atomic Number:** $55 \rightarrow 00110111$
- **Configuration:** s-block (Group 1) $\rightarrow 1000$
- **Valence Electrons:** $1 \rightarrow 001$
- **Electronegativity:** $0.79 < 2.5 \rightarrow 0$
- **Period:** $6 \rightarrow 110$
- **Group:** $1 \rightarrow 00001$
- **Vector:** 00110111 1000 001 0 110 00001

56. Barium (Ba)

- **Atomic Number:** $56 \rightarrow 00111000$
- **Configuration:** s-block (Group 2) $\rightarrow 1000$
- **Valence Electrons:** $2 \rightarrow 010$
- **Electronegativity:** $0.89 < 2.5 \rightarrow 0$
- **Period:** $6 \rightarrow 110$
- **Group:** $2 \rightarrow 00010$
- **Vector:** 00111000 1000 010 0 110 00010

57. Lanthanum (La)

- **Atomic Number:** 57 → 00111001
- **Configuration:** f-block (lanthanides) → 1111
- **Valence Electrons:** 2 ($6s^2$) → 010
- **Electronegativity:** $1.1 < 2.5$ → 0
- **Period:** 6 → 110
- **Group:** 3 → 00011
- **Vector:** 00111001 1111 010 0 110 00011

58. Cerium (Ce)

- **Atomic Number:** 58 → 00111010
- **Configuration:** f-block (lanthanides) → 1111
- **Valence Electrons:** 2 ($6s^2$) → 010
- **Electronegativity:** $1.12 < 2.5$ → 0
- **Period:** 6 → 110
- **Group:** 3 → 00011
- **Vector:** 00111010 1111 010 0 110 00011

59. Praseodymium (Pr)

- **Atomic Number:** 59 → 00111011
- **Configuration:** f-block (lanthanides) → 1111
- **Valence Electrons:** 2 ($6s^2$) → 010
- **Electronegativity:** $1.13 < 2.5$ → 0
- **Period:** 6 → 110
- **Group:** 3 → 00011
- **Vector:** 00111011 1111 010 0 110 00011

60. Neodymium (Nd)

- **Atomic Number:** 60 → 00111100
- **Configuration:** f-block (lanthanides) → 1111

- **Valence Electrons:** 2 ($6s^2$) $\rightarrow$ 010
- **Electronegativity:** 1.14 < 2.5 $\rightarrow$ 0
- **Period:** 6 $\rightarrow$ 110
- **Group:** 3 $\rightarrow$ 00011
- **Vector:** 00111100 1111 010 0 110 00011

61. Promethium (Pm)

- **Atomic Number:** 61 $\rightarrow$ 00111101
- **Configuration:** f-block (lanthanides) $\rightarrow$ 1111
- **Valence Electrons:** 2 ($6s^2$) $\rightarrow$ 010
- **Electronegativity:** 1.13 < 2.5 $\rightarrow$ 0
- **Period:** 6 $\rightarrow$ 110
- **Group:** 3 $\rightarrow$ 00011
- **Vector:** 00111101 1111 010 0 110 00011

62. Samarium (Sm)

- **Atomic Number:** 62 $\rightarrow$ 00111110
- **Configuration:** f-block (lanthanides) $\rightarrow$ 1111
- **Valence Electrons:** 2 ($6s^2$) $\rightarrow$ 010
- **Electronegativity:** 1.17 < 2.5 $\rightarrow$ 0
- **Period:** 6 $\rightarrow$ 110
- **Group:** 3 $\rightarrow$ 00011
- **Vector:** 00111110 1111 010 0 110 00011

63. Europium (Eu)

- **Atomic Number:** 63 $\rightarrow$ 00111111
- **Configuration:** f-block (lanthanides) $\rightarrow$ 1111
- **Valence Electrons:** 2 ($6s^2$) $\rightarrow$ 010
- **Electronegativity:** 1.2 < 2.5 $\rightarrow$ 0
- **Period:** 6 $\rightarrow$ 110
- **Group:** 3 $\rightarrow$ 00011
- **Vector:** 00111111 1111 010 0 110 00011

64. Gadolinium (Gd)

- **Atomic Number:** 64 → 01000000
- **Configuration:** f-block (lanthanides) → 1111
- **Valence Electrons:** 2 ($6s^2$) → 010
- **Electronegativity:** 1.2 < 2.5 → 0
- **Period:** 6 → 110
- **Group:** 3 → 00011
- **Vector:** 01000000 1111 010 0 110 00011

65. Terbium (Tb)

- **Atomic Number:** 65 → 01000001
- **Configuration:** f-block (lanthanides) → 1111
- **Valence Electrons:** 2 ($6s^2$) → 010
- **Electronegativity:** 1.2 < 2.5 → 0
- **Period:** 6 → 110
- **Group:** 3 → 00011
- **Vector:** 01000001 1111 010 0 110 00011

66. Dysprosium (Dy)

- **Atomic Number:** 66 → 01000010
- **Configuration:** f-block (lanthanides) → 1111
- **Valence Electrons:** 2 ($6s^2$) → 010
- **Electronegativity:** 1.22 < 2.5 → 0
- **Period:** 6 → 110
- **Group:** 3 → 00011
- **Vector:** 01000010 1111 010 0 110 00011

67. Holmium (Ho)

- **Atomic Number:** 67 → 01000011
- **Configuration:** f-block (lanthanides) → 1111

- **Valence Electrons:** 2 ($6s^2$) $\rightarrow$ 010
- **Electronegativity:** $1.23 < 2.5 \rightarrow 0$
- **Period:** 6 $\rightarrow$ 110
- **Group:** 3 $\rightarrow$ 00011
- **Vector:** 01000011 1111 010 0 110 00011

68. Erbium (Er)

- **Atomic Number:** 68 $\rightarrow$ 01000100
- **Configuration:** f-block (lanthanides) $\rightarrow$ 1111
- **Valence Electrons:** 2 ($6s^2$) $\rightarrow$ 010
- **Electronegativity:** $1.24 < 2.5 \rightarrow 0$
- **Period:** 6 $\rightarrow$ 110
- **Group:** 3 $\rightarrow$ 00011
- **Vector:** 01000100 1111 010 0 110 00011

69. Thulium (Tm)

- **Atomic Number:** 69 $\rightarrow$ 01000101
- **Configuration:** f-block (lanthanides) $\rightarrow$ 1111
- **Valence Electrons:** 2 ($6s^2$) $\rightarrow$ 010
- **Electronegativity:** $1.25 < 2.5 \rightarrow 0$
- **Period:** 6 $\rightarrow$ 110
- **Group:** 3 $\rightarrow$ 00011
- **Vector:** 01000101 1111 010 0 110 00011

70. Ytterbium (Yb)

- **Atomic Number:** 70 $\rightarrow$ 01000110
- **Configuration:** f-block (lanthanides) $\rightarrow$ 1111
- **Valence Electrons:** 2 ($6s^2$) $\rightarrow$ 010
- **Electronegativity:** $1.1 < 2.5 \rightarrow 0$
- **Period:** 6 $\rightarrow$ 110
- **Group:** 3 $\rightarrow$ 00011
- **Vector:** 01000110 1111 010 0 110 00011

71. Lutetium (Lu)

- **Atomic Number:** 71 → 01000111
- **Configuration:** f-block (lanthanides) → 1111
- **Valence Electrons:** 2 ($6s^2$) → 010
- **Electronegativity:** $1.27 < 2.5$ → 0
- **Period:** 6 → 110
- **Group:** 3 → 00011
- **Vector:** 01000111 1111 010 0 110 00011

72. Hafnium (Hf)

- **Atomic Number:** 72 → 01001000
- **Configuration:** d-block (Group 4) → 1010
- **Valence Electrons:** 2 ($6s^2$) → 010
- **Electronegativity:** $1.3 < 2.5$ → 0
- **Period:** 6 → 110
- **Group:** 4 → 00100
- **Vector:** 01001000 1010 010 0 110 00100

73. Tantalum (Ta)

- **Atomic Number:** 73 → 01001001
- **Configuration:** d-block (Group 5) → 1010
- **Valence Electrons:** 2 ($6s^2$) → 010
- **Electronegativity:** $1.5 < 2.5$ → 0
- **Period:** 6 → 110
- **Group:** 5 → 00101
- **Vector:** 01001001 1010 010 0 110 00101

74. Tungsten (W)

- **Atomic Number:** 74 → 01001010
- **Configuration:** d-block (Group 6) → 1010

- **Valence Electrons:** 2 ($6s^2$) $\rightarrow$ 010
- **Electronegativity:** $2.36 < 2.5 \rightarrow 0$
- **Period:** 6 $\rightarrow$ 110
- **Group:** 6 $\rightarrow$ 00110
- **Vector:** 01001010 1010 010 0 110 00110

75. Rhenium (Re)

- **Atomic Number:** 75 $\rightarrow$ 01001011
- **Configuration:** d-block (Group 7) $\rightarrow$ 1010
- **Valence Electrons:** 2 ($6s^2$) $\rightarrow$ 010
- **Electronegativity:** $1.9 < 2.5 \rightarrow 0$
- **Period:** 6 $\rightarrow$ 110
- **Group:** 7 $\rightarrow$ 00111
- **Vector:** 01001011 1010 010 0 110 00111

76. Osmium (Os)

- **Atomic Number:** 76 $\rightarrow$ 01001100
- **Configuration:** d-block (Group 8) $\rightarrow$ 1010
- **Valence Electrons:** 2 ($6s^2$) $\rightarrow$ 010
- **Electronegativity:** $2.2 < 2.5 \rightarrow 0$
- **Period:** 6 $\rightarrow$ 110
- **Group:** 8 $\rightarrow$ 01000
- **Vector:** 01001100 1010 010 0 110 01000

77. Iridium (Ir)

- **Atomic Number:** 77 $\rightarrow$ 01001101
- **Configuration:** d-block (Group 9) $\rightarrow$ 1010
- **Valence Electrons:** 2 ($6s^2$) $\rightarrow$ 010
- **Electronegativity:** $2.2 < 2.5 \rightarrow 0$
- **Period:** 6 $\rightarrow$ 110
- **Group:** 9 $\rightarrow$ 01001
- **Vector:** 01001101 1010 010 0 110 01001

78. Platinum (Pt)

- **Atomic Number:** 78 $\rightarrow$ 01001110
- **Configuration:** d-block (Group 10) $\rightarrow$ 1010
- **Valence Electrons:** 1 ($6s^1$ due to exception) $\rightarrow$ 001
- **Electronegativity:** $2.28 < 2.5 \rightarrow 0$
- **Period:** 6 $\rightarrow$ 110
- **Group:** 10 $\rightarrow$ 01010
- **Vector:** 01001110 1010 001 0 110 01010

79. Gold (Au)

- **Atomic Number:** 79 $\rightarrow$ 01001111
- **Configuration:** d-block (Group 11) $\rightarrow$ 1010
- **Valence Electrons:** 1 ($6s^1$) $\rightarrow$ 001
- **Electronegativity:** $2.54 \geq 2.5 \rightarrow 1$
- **Period:** 6 $\rightarrow$ 110
- **Group:** 11 $\rightarrow$ 01011
- **Vector:** 01001111 1010 001 1 110 01011

80. Mercury (Hg)

- **Atomic Number:** 80 $\rightarrow$ 01010000
- **Configuration:** d-block (Group 12) $\rightarrow$ 1010
- **Valence Electrons:** 2 ($6s^2$) $\rightarrow$ 010
- **Electronegativity:** $2.0 < 2.5 \rightarrow 0$
- **Period:** 6 $\rightarrow$ 110
- **Group:** 12 $\rightarrow$ 01100
- **Vector:** 01010000 1010 010 0 110 01100

81. Thallium (Tl)

- **Atomic Number:** 81 $\rightarrow$ 01010001
- **Configuration:** p-block (Group 13) $\rightarrow$ 1100

- **Valence Electrons:** 3 → 011
- **Electronegativity:** 1.62 < 2.5 → 0
- **Period:** 6 → 110
- **Group:** 13 → 01101
- **Vector:** 01010001 1100 011 0 110 01101

82. Lead (Pb)

- **Atomic Number:** 82 → 01010010
- **Configuration:** p-block (Group 14) → 1100
- **Valence Electrons:** 4 → 100
- **Electronegativity:** 2.33 < 2.5 → 0
- **Period:** 6 → 110
- **Group:** 14 → 01110
- **Vector:** 01010010 1100 100 0 110 01110

83. Bismuth (Bi)

- **Atomic Number:** 83 → 01010011
- **Configuration:** p-block (Group 15) → 1100
- **Valence Electrons:** 5 → 101
- **Electronegativity:** 2.02 < 2.5 → 0
- **Period:** 6 → 110
- **Group:** 15 → 01111
- **Vector:** 01010011 1100 101 0 110 01111

84. Polonium (Po)

- **Atomic Number:** 84 → 01010100
- **Configuration:** p-block (Group 16) → 1100
- **Valence Electrons:** 6 → 110
- **Electronegativity:** 2.0 < 2.5 → 0
- **Period:** 6 → 110
- **Group:** 16 → 10000
- **Vector:** 01010100 1100 110 0 110 10000

85. Astatine (At)

- **Atomic Number:** 85 → 01010101
- **Configuration:** p-block (Group 17) → 1100
- **Valence Electrons:** 7 → 111
- **Electronegativity:** 2.2 < 2.5 → 0
- **Period:** 6 → 110
- **Group:** 17 → 10001
- **Vector:** 01010101 1100 111 0 110 10001

86. Radon (Rn)

- **Atomic Number:** 86 → 01010110
- **Configuration:** p-block (Group 18) → 1100
- **Valence Electrons:** Noble gas, full shell → 000
- **Electronegativity:** Undefined, set to 0 → 0
- **Period:** 6 → 110
- **Group:** 18 → 10010
- **Vector:** 01010110 1100 000 0 110 10010

87. Francium (Fr)

- **Atomic Number:** 87 → 01010111
- **Configuration:** s-block (Group 1) → 1000
- **Valence Electrons:** 1 → 001
- **Electronegativity:** 0.7 < 2.5 → 0
- **Period:** 7 → 111
- **Group:** 1 → 00001
- **Vector:** 01010111 1000 001 0 111 00001

88. Radium (Ra)

- **Atomic Number:** 88 → 01011000
- **Configuration:** s-block (Group 2) → 1000

- **Valence Electrons:** $2 \rightarrow 010$
- **Electronegativity:** $0.9 < 2.5 \rightarrow 0$
- **Period:** $7 \rightarrow 111$
- **Group:** $2 \rightarrow 00010$
- **Vector:** 01011000 1000 010 0 111 00010

89. Actinium (Ac)

- **Atomic Number:** 89 $\rightarrow 01011001$
- **Configuration:** f-block (actinides) $\rightarrow 1111$
- **Valence Electrons:** 2 ($7s^2$) $\rightarrow 010$
- **Electronegativity:** $1.1 < 2.5 \rightarrow 0$
- **Period:** $7 \rightarrow 111$
- **Group:** 3 $\rightarrow 00011$
- **Vector:** 01011001 1111 010 0 111 00011

90. Thorium (Th)

- **Atomic Number:** 90 $\rightarrow 01011010$
- **Configuration:** f-block (actinides) $\rightarrow 1111$
- **Valence Electrons:** 2 ($7s^2$) $\rightarrow 010$
- **Electronegativity:** $1.3 < 2.5 \rightarrow 0$
- **Period:** $7 \rightarrow 111$
- **Group:** 3 $\rightarrow 00011$
- **Vector:** 01011010 1111 010 0 111 00011

91. Protactinium (Pa)

- **Atomic Number:** 91 $\rightarrow 01011011$
- **Configuration:** f-block (actinides) $\rightarrow 1111$
- **Valence Electrons:** 2 ($7s^2$) $\rightarrow 010$
- **Electronegativity:** $1.5 < 2.5 \rightarrow 0$
- **Period:** $7 \rightarrow 111$
- **Group:** 3 $\rightarrow 00011$
- **Vector:** 01011011 1111 010 0 111 00011

92. Uranium (U)

- **Atomic Number:** 92 → 01011100
- **Configuration:** f-block (actinides) → 1111
- **Valence Electrons:** 2 ($7s^2$) → 010
- **Electronegativity:** $1.38 < 2.5$ → 0
- **Period:** 7 → 111
- **Group:** 3 → 00011
- **Vector:** 01011100 1111 010 0 111 00011

93. Neptunium (Np)

- **Atomic Number:** 93 → 01011101
- **Configuration:** f-block (actinides) → 1111
- **Valence Electrons:** 2 ($7s^2$) → 010
- **Electronegativity:** $1.36 < 2.5$ → 0
- **Period:** 7 → 111
- **Group:** 3 → 00011
- **Vector:** 01011101 1111 010 0 111 00011

94. Plutonium (Pu)

- **Atomic Number:** 94 → 01011110
- **Configuration:** f-block (actinides) → 1111
- **Valence Electrons:** 2 ($7s^2$) → 010
- **Electronegativity:** $1.28 < 2.5$ → 0
- **Period:** 7 → 111
- **Group:** 3 → 00011
- **Vector:** 01011110 1111 010 0 111 00011

95. Americium (Am)

- **Atomic Number:** 95 → 01011111
- **Configuration:** f-block (actinides) → 1111

- **Valence Electrons:** 2 ($7s^2$) $\rightarrow$ 010
- **Electronegativity:** 1.3 < 2.5 $\rightarrow$ 0
- **Period:** 7 $\rightarrow$ 111
- **Group:** 3 $\rightarrow$ 00011
- **Vector:** 01011111 1111 010 0 111 00011

96. Curium (Cm)

- **Atomic Number:** 96 $\rightarrow$ 01100000
- **Configuration:** f-block (actinides) $\rightarrow$ 1111
- **Valence Electrons:** 2 ($7s^2$) $\rightarrow$ 010
- **Electronegativity:** 1.3 < 2.5 $\rightarrow$ 0
- **Period:** 7 $\rightarrow$ 111
- **Group:** 3 $\rightarrow$ 00011
- **Vector:** 01100000 1111 010 0 111 00011

97. Berkelium (Bk)

- **Atomic Number:** 97 $\rightarrow$ 01100001
- **Configuration:** f-block (actinides) $\rightarrow$ 1111
- **Valence Electrons:** 2 ($7s^2$) $\rightarrow$ 010
- **Electronegativity:** 1.3 < 2.5 $\rightarrow$ 0
- **Period:** 7 $\rightarrow$ 111
- **Group:** 3 $\rightarrow$ 00011
- **Vector:** 01100001 1111 010 0 111 00011

98. Californium (Cf)

- **Atomic Number:** 98 $\rightarrow$ 01100010
- **Configuration:** f-block (actinides) $\rightarrow$ 1111
- **Valence Electrons:** 2 ($7s^2$) $\rightarrow$ 010
- **Electronegativity:** 1.3 < 2.5 $\rightarrow$ 0
- **Period:** 7 $\rightarrow$ 111
- **Group:** 3 $\rightarrow$ 00011
- **Vector:** 01100010 1111 010 0 111 00011

99. Einsteinium (Es)

- **Atomic Number:** 99 → 01100011
- **Configuration:** f-block (actinides) → 1111
- **Valence Electrons:** 2 ($7s^2$) → 010
- **Electronegativity:** $1.3 < 2.5$ → 0
- **Period:** 7 → 111
- **Group:** 3 → 00011
- **Vector:** 01100011 1111 010 0 111 00011

100. Fermium (Fm)

- **Atomic Number:** 100 → 01100100
- **Configuration:** f-block (actinides) → 1111
- **Valence Electrons:** 2 ($7s^2$) → 010
- **Electronegativity:** $1.3 < 2.5$ → 0
- **Period:** 7 → 111
- **Group:** 3 → 00011
- **Vector:** 01100100 1111 010 0 111 00011

101. Mendelevium (Md)

- **Atomic Number:** 101 → 01100101
- **Configuration:** f-block (actinides) → 1111
- **Valence Electrons:** 2 ($7s^2$) → 010
- **Electronegativity:** $1.3 < 2.5$ → 0
- **Period:** 7 → 111
- **Group:** 3 → 00011
- **Vector:** 01100101 1111 010 0 111 00011

102. Nobelium (No)

- **Atomic Number:** 102 → 01100110
- **Configuration:** f-block (actinides) → 1111

- **Valence Electrons:** 2 ($7s^2$) $\rightarrow$ 010
- **Electronegativity:** $1.3 < 2.5 \rightarrow 0$
- **Period:** 7 $\rightarrow$ 111
- **Group:** 3 $\rightarrow$ 00011
- **Vector:** 01100110 1111 010 0 111 00011

103. Lawrencium (Lr)

- **Atomic Number:** 103 $\rightarrow$ 01100111
- **Configuration:** f-block (actinides) $\rightarrow$ 1111
- **Valence Electrons:** 2 ($7s^2$) $\rightarrow$ 010
- **Electronegativity:** $1.3 < 2.5 \rightarrow 0$
- **Period:** 7 $\rightarrow$ 111
- **Group:** 3 $\rightarrow$ 00011
- **Vector:** 01100111 1111 010 0 111 00011

104. Rutherfordium (Rf)

- **Atomic Number:** 104 $\rightarrow$ 01101000
- **Configuration:** d-block (Group 4) $\rightarrow$ 1010
- **Valence Electrons:** 2 ($7s^2$) $\rightarrow$ 010
- **Electronegativity:** Estimated $\sim 1.3 < 2.5 \rightarrow 0$
- **Period:** 7 $\rightarrow$ 111
- **Group:** 4 $\rightarrow$ 00100
- **Vector:** 01101000 1010 010 0 111 00100

105. Dubnium (Db)

- **Atomic Number:** 105 $\rightarrow$ 01101001
- **Configuration:** d-block (Group 5) $\rightarrow$ 1010
- **Valence Electrons:** 2 ($7s^2$) $\rightarrow$ 010
- **Electronegativity:** Estimated $\sim 1.3 < 2.5 \rightarrow 0$
- **Period:** 7 $\rightarrow$ 111
- **Group:** 5 $\rightarrow$ 00101
- **Vector:** 01101001 1010 010 0 111 00101

106. Seaborgium (Sg)

- **Atomic Number:** 106 → 01101010
- **Configuration:** d-block (Group 6) → 1010
- **Valence Electrons:** 2 ($7s^2$) → 010
- **Electronegativity:** Estimated $\sim 1.3 < 2.5 \rightarrow 0$
- **Period:** 7 → 111
- **Group:** 6 → 00110
- **Vector:** 01101010 1010 010 0 111 00110

107. Bohrium (Bh)

- **Atomic Number:** 107 → 01101011
- **Configuration:** d-block (Group 7) → 1010
- **Valence Electrons:** 2 ($7s^2$) → 010
- **Electronegativity:** Estimated $\sim 1.3 < 2.5 \rightarrow 0$
- **Period:** 7 → 111
- **Group:** 7 → 00111
- **Vector:** 01101011 1010 010 0 111 00111

108. Hassium (Hs)

- **Atomic Number:** 108 → 01101100
- **Configuration:** d-block (Group 8) → 1010
- **Valence Electrons:** 2 ($7s^2$) → 010
- **Electronegativity:** Estimated $\sim 1.3 < 2.5 \rightarrow 0$
- **Period:** 7 → 111
- **Group:** 8 → 01000
- **Vector:** 01101100 1010 010 0 111 01000

109. Meitnerium (Mt)

- **Atomic Number:** 109 → 01101101
- **Configuration:** d-block (Group 9) → 1010

- **Valence Electrons:** 2 ($7s^2$) $\rightarrow$ 010
- **Electronegativity:** Estimated $\sim 1.3 < 2.5 \rightarrow 0$
- **Period:** 7 $\rightarrow$ 111
- **Group:** 9 $\rightarrow$ 01001
- **Vector:** 01101101 1010 010 0 111 01001

110. Darmstadtium (Ds)

- **Atomic Number:** 110 $\rightarrow$ 01101110
- **Configuration:** d-block (Group 10) $\rightarrow$ 1010
- **Valence Electrons:** 2 ($7s^2$) $\rightarrow$ 010
- **Electronegativity:** Estimated $\sim 1.3 < 2.5 \rightarrow 0$
- **Period:** 7 $\rightarrow$ 111
- **Group:** 10 $\rightarrow$ 01010
- **Vector:** 01101110 1010 010 0 111 01010

111. Roentgenium (Rg)

- **Atomic Number:** 111 $\rightarrow$ 01101111
- **Configuration:** d-block (Group 11) $\rightarrow$ 1010
- **Valence Electrons:** 2 ($7s^2$) $\rightarrow$ 010
- **Electronegativity:** Estimated $\sim 1.3 < 2.5 \rightarrow 0$
- **Period:** 7 $\rightarrow$ 111
- **Group:** 11 $\rightarrow$ 01011
- **Vector:** 01101111 1010 010 0 111 01011

112. Copernicium (Cn)

- **Atomic Number:** 112 $\rightarrow$ 01110000
- **Configuration:** d-block (Group 12) $\rightarrow$ 1010
- **Valence Electrons:** 2 ($7s^2$) $\rightarrow$ 010
- **Electronegativity:** Estimated $\sim 1.3 < 2.5 \rightarrow 0$
- **Period:** 7 $\rightarrow$ 111
- **Group:** 12 $\rightarrow$ 01100
- **Vector:** 01110000 1010 010 0 111 01100

113. Nihonium (Nh)

- **Atomic Number:** 113 $\rightarrow$ 01110001
- **Configuration:** p-block (Group 13) $\rightarrow$ 1100
- **Valence Electrons:** 3 $\rightarrow$ 011
- **Electronegativity:** Estimated $\sim 1.3 < 2.5 \rightarrow 0$
- **Period:** 7 $\rightarrow$ 111
- **Group:** 13 $\rightarrow$ 01101
- **Vector:** 01110001 1100 011 0 111 01101

114. Flerovium (Fl)

- **Atomic Number:** 114 $\rightarrow$ 01110010
- **Configuration:** p-block (Group 14) $\rightarrow$ 1100
- **Valence Electrons:** 4 $\rightarrow$ 100
- **Electronegativity:** Estimated $\sim 1.3 < 2.5 \rightarrow 0$
- **Period:** 7 $\rightarrow$ 111
- **Group:** 14 $\rightarrow$ 01110
- **Vector:** 01110010 1100 100 0 111 01110

115. Moscovium (Mc)

- **Atomic Number:** 115 $\rightarrow$ 01110011
- **Configuration:** p-block (Group 15) $\rightarrow$ 1100
- **Valence Electrons:** 5 $\rightarrow$ 101
- **Electronegativity:** Estimated $\sim 1.5 < 2.5 \rightarrow 0$
- **Period:** 7 $\rightarrow$ 111
- **Group:** 15 $\rightarrow$ 01111
- **Vector:** 01110011 1100 101 0 111 01111

116. Livermorium (Lv)

- **Atomic Number:** 116 $\rightarrow$ 01110100
- **Configuration:** p-block (Group 16) $\rightarrow$ 1100

- **Valence Electrons:** 6 → 110
- **Electronegativity:** Estimated $\sim 1.5 < 2.5 \rightarrow 0$
- **Period:** 7 → 111
- **Group:** 16 → 10000
- **Vector:** 01110100 1100 110 0 111 10000

117. Tennessine (Ts)

- **Atomic Number:** 117 → 01110101
- **Configuration:** p-block (Group 17) → 1100
- **Valence Electrons:** 7 → 111
- **Electronegativity:** Estimated $\sim 2.1 < 2.5 \rightarrow 0$
- **Period:** 7 → 111
- **Group:** 17 → 10001
- **Vector:** 01110101 1100 111 0 111 10001

118. Oganesson (Og)

- **Atomic Number:** 118 → 01110110
- **Configuration:** p-block (Group 18) → 1100
- **Valence Electrons:** Noble gas, full shell → 000
- **Electronegativity:** Estimated $\sim 2.0 < 2.5 \rightarrow 0$
- **Period:** 7 → 111
- **Group:** 18 → 10010
- **Vector:** 01110110 1100 000 0 111 10010